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## Fourier transform infrared (FTIR) spectroscopy gel electrolyte studies of chitosan polymer / Nurul A'iin Morsid.

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FOURIER TRANSFORM INFRARED SPECTROSCOPY (FTIR)  
STUDIES OF CHITOSAN POLYMER  
GEL ELECTROLYTES

By

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Research Report submitted in partial fulfillment of  
the requirement for the degree of  
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Department of Physical Sciences  
Faculty of Science and Technology  
UNIVERSITY MALAYSIA TERENGGANU  
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## PENGAKUAN DAN PENGESAHAN LAPORAN PROJEK PENYELIDIKAN I DAN II

Adalah ini diakui dan disahkan bahawa laporan penyelidikan bertajuk:

FOURIER TRANSFORM INFRARED (FTIR) SPECTROSCOPY STUDIES OF CHITOSAN POLYMER GEL ELECTROLYTES oleh NURRUL A'IIN MORSID, no matrik UK9331 telah diperiksa dan semua pembetulan yang disarankan telah dilakukan. Laporan ini dikemukakan kepada Jabatan Sains Fizik sebagai memenuhi sebahagian daripada keperluan Ijazah Sarjana Muda Sains Gunaan (Fizik Elektronik dan Instrumentasi), Fakulti Sains dan Teknologi, Universiti Malaysia Terengganu.

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## CONTENTS

Title page	i
Form of confirmation thesis	ii
Acknowledgements	iii
Table of contents	iv
List of figures	vi
List of tables	viii
List of Abbreviations	xi
Abstract	ix
Abstrak	x

### **CHAPTER 1: INTRODUCTION**

### **CHAPTER 2: LITERATURE REVIEW**

2.1 Polymer Electrolytes	4
2.2 Gel Electrolytes	5
2.3 Fourier Transform Infrared Spectroscopy (FTIR)	6
2.4 Electrochemical Impedance Spectroscopy (EIS)	7

### **CHAPTER 3: METHODOLOGY**

3.1 Sample Preparation	9
3.2 Sample Characterization	
3.2.1 Fourier Transform Infrared Spectroscopy (FTIR) Analysis	10
3.2.2 Electrochemical Impedance Spectroscopy (EIS) Analysis	12

## CHAPTER 4: RESULTS AND DISCUSSIONS

4.1	Sample preparation	15
4.2	FTIR for Chitosan Acetate (CA)	15
4.3	FTIR for Ethylene Carbonate (EC)	16
4.4	FTIR for Propylene Carbonate (PC)	17
4.5	FTIR for Poly (methyl methacrylate) (PMMA)	19
4.6	FTIR for Salicylic Acid (SA)	20
4.7	FTIR of CA-EC-PC-PMMA-SA 5 wt.%, 7 wt.%, 10 wt.%, 15 wt.%	22
4.8	Summary of FTIR studies	27
4.9	Electrical Impedance Spectroscopy (EIS) studies	27

## CHAPTER 5: CONCLUSION AND SUGGESTIONS FOR FURTHER WORK

## REFERENCES

## CURRICULUM VITAE

## LIST OF FIGURES

<b><i>Figure</i></b>	<b><i>Captions</i></b>	<b><i>Page</i></b>
Figure 2.1	Structural formula of Chitosan	5
Figure 3.1	Perkin-Elmer System 2000 FTIR Spectrometer	10
Figure 3.2	HIOKI 3531 – LCR Hi-Tester	12
Figure 3.3	Sample holder	13
Figure 4.1	Structural formula of CA	15
Figure 4.2	ATR-FTIR spectrum of CA	16
Figure 4.3	Structure of EC	16
Figure 4.4	ATR-FTIR spectrum of EC	17
Figure 4.5	Structure of PC	18
Figure 4.6	ATR-FTIR spectrum of PC	18
Figure 4.7	Structural formula of PMMA	19
Figure 4.8	ATR-FTIR spectrum of PMMA	19
Figure 4.9	Structure of SA	20
Figure 4.10	ATR-FTIR spectrum of SA	20
Figure 4.11	The FTIR spectra of (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA (IV) 10wt.% SA (V) 15wt.% SA	22
Figure 4.12(a)	Parts of the chitosan-SA in the wavenumber region between 1490 – 1700 cm <sup>-1</sup> for (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA (IV) 10wt.% SA (V) 15wt.% SA	23
Figure 4.12(b)	Parts of the chitosan-SA in the wavenumber region between 1500 – 1650 cm <sup>-1</sup> for (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA (IV) 10wt.% SA (V) 15wt.% SA	23
Figure 4.13(a)	Parts of the chitosan-SA in the region between 1350 – 1470 cm <sup>-1</sup> for (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA (IV) 10wt.% SA (V) 15wt.% SA	25
Figure 4.13(b)	Parts of the chitosan-SA in the region between 660 – 800 cm <sup>-1</sup> for (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA (IV) 10wt.% SA (V) 15wt.% SA	25
Figure 4.14(a)	Parts of the chitosan-SA in the region between 2760 – 3070 cm <sup>-1</sup> for (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA (IV) 10wt.% SA (V) 15wt.% SA	26

Figure 4.14(b) Parts of the chitosan-SA in the region between 1090 – 1210 cm <sup>-1</sup>	26
for (I) pure CA and CA with (II) 5wt.% SA (III) 7wt.% SA	
(IV) 10wt.% SA (V) 15wt.% SA	
Figure 4.15(a) Impedance plots of CA	29
Figure 4.15(b) Impedance plots of chitosan with 5wt.% SA	29
Figure 4.15(c) Impedance plots of chitosan with 7wt.% SA	29
Figure 4.15(d) Impedance plots of chitosan with 10wt.% SA	29
Figure 4.15(e) Impedance plots of chitosan with 10wt.% SA	29

## LIST OF TABLES

<b><i>Tables</i></b>		<b><i>Page</i></b>
Table 3.1	The quantity of the sample for preparing gel electrolyte	10
Table 4.1	The conductivity of polymer gel electrolytes	30

## LIST OF ABBREVIATIONS

ATR	Attenuated Total Reflection
CA	Chitosan acetate
C <sub>φ</sub>	Carbon of the aromatic ring
DBP	Dibutyl phthalate
DMF	Dimethyl formamide
EC	Ethylene carbonate
EIS	Electrochemical Impedance Spectroscopy
FTIR	Fourier Transform Infrared Spectroscopy
GPEs	Gel polymer electrolytes
PAN	Polyacrylonitrile
R <sub>b</sub>	Bulk resistance
PC	Propylene carbonate
PCS	Photon Correlation Spectroscopy
PEO	Polyethylene oxide
PMMA	Poly(methyl methacrylate)
SA	Salicylic acid
SEM	Scanning Electron Microscopy
UV	Ultra violet
v <sub>s</sub>	Symmetric stretching vibration
v <sub>as</sub>	Asymmetric stretching vibration
ZnSe	Zinc Selanium
Z'	Real impedance
Z''	Imaginary impedance
σ	Conductivity
δ	Bending mode
ω	Wagging mode
ρ	Rocking mode
τ	Twisting mode

# ABSTRACT

**ABSTRACT**

In this work, the synthesis of proton-conducting gel membranes obtained by incorporating in organic acid in poly(methyl methacrylate) (PMMA) matrix. Propylene carbonate (PC) and ethylene carbonate (EC) were used as solvents. Organic acid used in this work is salicylic acid (SA). The ionic conductivity of the electrolyte is  $2.65 \times 10^{-5}$  S cm<sup>-1</sup> at room temperature. The gels were characterized by Fourier Transform Infrared (FTIR) Spectroscopy using Perkin-Elmer System 2000 FTIR Spectrometer with Attenuated Total Reflection (ATR) accessories. Observation on FTIR spectrum of the chitosan-salt gels is at 1389 cm<sup>-1</sup> due to vibration mode of carboxyl group,  $\nu(\text{COO}^-)$  which the salicylic acid proton donor has protonated. This proved that H<sup>+</sup> ions have dissociated as proton conductor. The obtained spectroscopic data has been correlated with the conductivity performance of chitosan-salt based polymer electrolytes.

ABSTRAK

**ABSTRAK**

Dalam kajian ini, gabungan gel membrane pengkonduksi-proton diperolehi dengan mencampurkan asid organik bersama matriks polimer “poly(methyl methacrylate)” (PMMA). Propilena karbonat (PC) dan etilena karbonat (EC) pula digunakan sebagai pelarut. Asid organik yang digunakan dalam kajian ini ialah asid salisilik (SA). Ion kekonduksian bagi elektrolit pada suhu bilik ialah  $2.65 \times 10^{-5} \text{ S cm}^{-1}$ . Sifat-sifat bagi suatu gel dikenalpasti dengan menggunakan “Fourier Transform Infrared (FTIR) Spectroscopy” jenis Perkin-Elmer 2000 FTIR Spektrometer bersama tambahan sistem iaitu “Attenuated Total reflection” (ATR). Hasil pemerhatian ke atas spektrum oleh FTIR bagi gel chitosan-SA, didapati pada  $1389 \text{ cm}^{-1}$  menujukkan mod getaran bagi kumpulan karboksil ( $\text{COO}^-$ ) di mana proton penderma asid salisilik (SA) terurai. Ini membuktikan bahawa ion  $\text{H}^+$  telah mengurai sebagai konduktor proton. Data spektroskopik yang diperolehi adalah berhubung kait dengan nilai kekonduksian bagi polimer elektrolit chitosan-SA.