

LABORATORY PREPARATION AND ANALYSIS OF
A PHOSPHATIC FERTILIZERS

S. PRABHAKARAN A/L SHANMUGAM

FACULTY OF SCIENCE AND TECHNOLOGY
KOLEJ UNIVERSITI TERENGGANU
UNIVERSITI PUTRA MALAYSIA

LP
13
FST
5
2001

2001

1100024738

CUW 1313

LP 13 FST 5 2001



1100024738

Laboratory preparation and analysis of a phosphatic fertilizers /
S. Prabhakaran a/l Shanmugan.

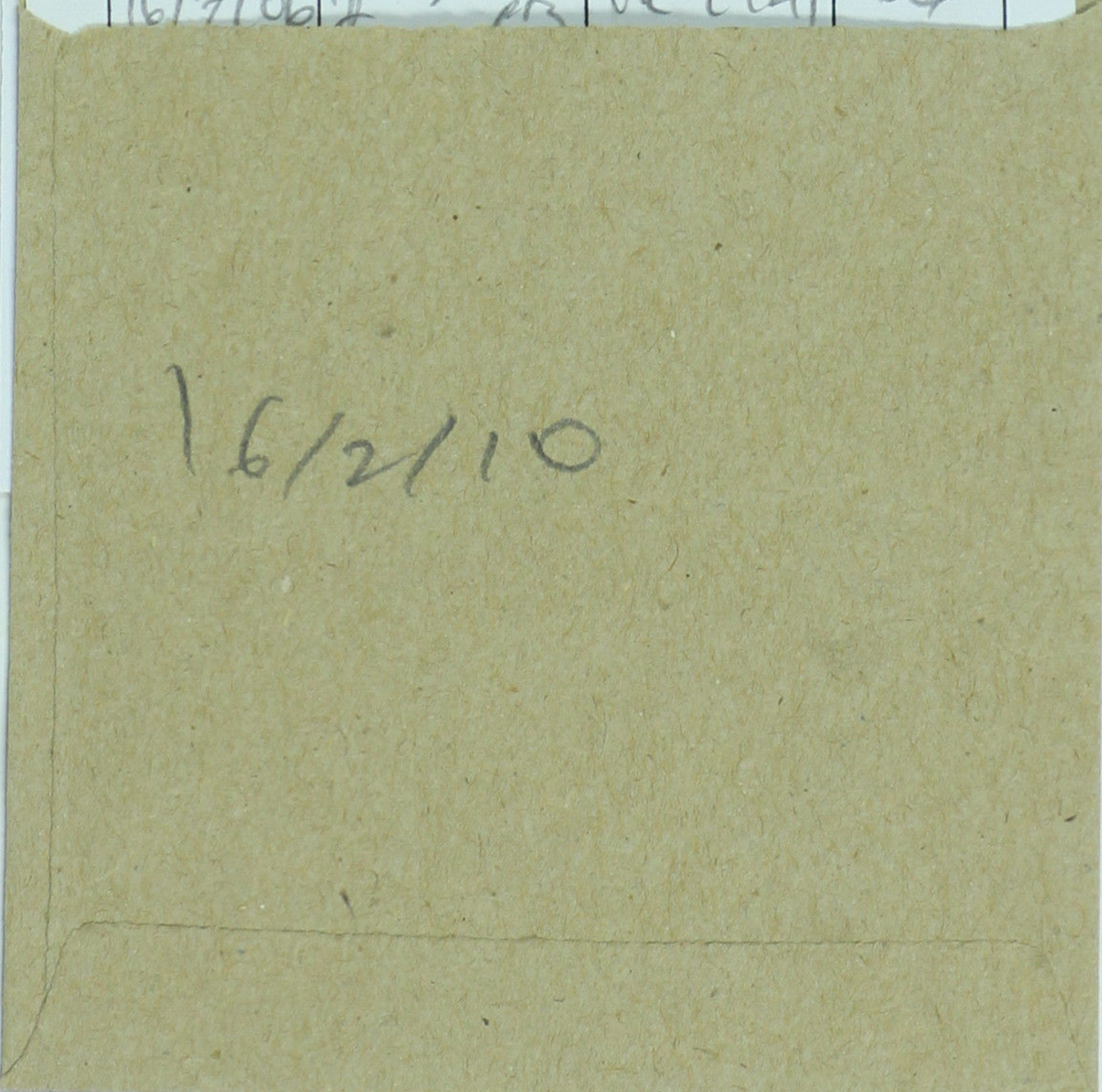


PERPUSTAKAAN
KOLEJ UNIVERSITI SAINS & TEKNOLOGI MALAYSIA
21030 KUALA TERENGGANU

1100024738		

1100024738

PERPUSTAKAAN KOLEJ UNIVERSITI SAINS & TEKNOLOGI MALAYSIA (KUSTEM)			
Pengarang S. Prabhakaran / Shanmugan		No. Panggilan HP PST	
Judul Laboratory preparation and analysis of			
Tarikh	Waktu Pemulangan	Nombor Ahli	Tanda tangan
7/7/03		UM 5867	
16/7/06	2 30	UK 2941	



16/2/10

LP
13
FST
5
2001

HAK MILIK
PERPUSTAKAAN KUSTEM

LABORATORY PREPARATION AND ANALYSIS OF A PHOSPHATIC FERTILIZERS

S.PRABHAKARAN A/L SHANMUGAM

PUSAT PEMBELAJARAN DIGITAL SULTANAH NUR FAHIMAH

FACULTY OF SCIENCE AND TECHNOLOGY
KOLEJ UNIVERSITI TERENGGANU
UNIVERSITI PUTRA MALAYSIA

2001

1100024738

LABORATORY PREPARATION AND ANALYSIS OF A PHOSPHATIC
FERTILIZERS

BY

S.PRABHAKARAN A/L SHANMUGAM

Thesis is submitted in partial fulfillment of the requirements for the Degree
of Science (Hons)

FACULTY OF SCIENCE AND TECHNOLOGY
KOLEJ UNIVERSITI TERENGGANU
UNIVERSITY PUTRA MALAYSIA

2001

LABORATORY PREPARATION AND ANALYSIS OF A PHOSPHATIC
FERTILIZERS

BY

S.PRABHAKARAN A/L SHANMUGAM


Approved By:

Supervisor


(Dr. Mohd Misbahul Amin)

Date: 25.03.2001

Coordinator


(Puan Marinah Mohd Ariffin)

Date: 27/03/01

Head of Chemistry


(Assoc. Prof. Dr. Ku Halim Ku Bulat)

Date: 27 Mar 2001

ACKNOWLEDGEMENT

I would like to thank my supervisor, Dr. Mohd Misbahul Amin, in assisting me with the final year project and for his patience in guiding me throughout the project.

I would also like to thank, head of chemistry and also my academic advisor, Assoc. Prof. Dr. Ku Halim Ku Bulat, for his guidance and help during the course of the project. My thanks to Professor Dr. Hamdan Suhaimi , for his help, invaluable advise, consideration and allowing me to carry out the study at UPMT.

My sincere appreciation is extended to all the lecturers and staffs of FST especially lab assistances Mizi, Kak Habsah and Man, for their support and encouragement during my Degree studies here in UPMT. Not forgetting my course mates, lab mates and housemates, Ah Teck, Ah Siang, Soo Hang and Ang, for their care and helpfulness who makes these years of study a most memorable one.

My thanks to my seniors, who helped me to adapt myself here in UPMT, Cikgu Thamo and wife, Anthony, Segar, and Rajeni for respecting and taking care of me. My thanks also goes out to all my friends for the advice and fun we had.

My heartiest thanks and appreciation to my 3 wonderful close friends; Devan, Santhi , and Viola , for the extraordinary and exciting time we had together.

Last but not least, I also feel indebted to my parents ,Mr and Mrs Shanmugam, my brothers, Narendran and Sarkuna, for their guidance, constant support and encouragement during the course of my studies in UPM Terengganu (UPMT), during the good and bad times we went through together.

ABSTRAK

Monokalsium fosfat, trikalsium fosfat, nitric fosfat dan diammonium fosfat merupakan baja yang biasa digunakan dalam pertanian sebagai baja fosfat. Oleh kerana, baja-baja ini berbeza dari segi kimia, maka adalah penting dibuat perbandingan dari segi keupayaan baja-baja tersebut membekalkan kandungan fosforus, P dalam bentuk fosforus pentaoksida (P_2O_5). Jadi, tujuan kajian ini dijalankan adalah untuk menghasilkan baja-baja fosfat dalam makmal dengan kaedah yang mudah. Proses untuk menghasilkan baja monokalsium fosfat adalah dengan mencampurkan serbuk fosfat batu dengan asid sulfurik dan untuk trikalsium fosfat pula, proses adalah dengan mencampurkan serbuk fosfat batu dengan asid fosforik. Baja yang dihasilkan dengan cara ini adalah murah dari segi nilai pertanian. Diammonium fosfat boleh dihasilkan dengan mencampurkan ammonia asid fosforik. Bagi fosfat nitrik, ia boleh dihasilkan dengan mencampurkan serbuk fosfat batu dengan asid nitrik. Keputusan dianalisa dengan menggunakan biasan X-ray. Ia untuk mengesahkan bahawa sampel-sampel mengandungi P_2O_5 dalam setiap baja yang dihasilkan. Monokalsium fosfat, trikalsium fosfat dan diammonium fosfat boleh dihasilkan dalam makmal kecuali fosfat nitrik. Nilai kualiti pertanian bagi baja-baja fosfat banyak bergantung kepada kuantiti, P, dalam bentuk P_2O_5 .

ABSTRACT

Single superphosphate (SSP), triple superphosphate (TSP), nitric phosphate and diammonium phosphate (DAP) are the most commonly used phosphate fertilizers. Since they differ in some chemical aspects it is important to compare their ability to supply phosphorus, P, in the form of phosphorous pentoxide (P_2O_5). The objective of this research was to manufacture phosphate fertilizer in simple laboratory method or at laboratory scale. The process to manufacture SSP consist of mixing powdered phosphate rock with sulphuric acid and for TSP, the process consist of mixing powered phosphate rock with phosphoric acid. This form of low cost fertilizer manufactured is suitable for improving the agronomic value. Diammonium phosphate is manufactured using ammonia and phosphoric acid. As for nitric acid, powdered phosphate rock is reacted with nitric acid. The result was analyzed using X-ray diffraction analysis. This is to confirmed that the samples governs the available P_2O_5 content in each phosphatic fertilizer. SSP, TSP and DAP can be manufactured using simple laboratory method except for nitric phosphate. The agronomic value of phosphate fertilizer highly depends upon the amount of P in P_2O_5 available in the fertilizer.