

ASSOCIATION BEHAVIOUR OF
MIXED BILE SALT AND SODIUM OCTANOATE
IN AQUEOUS SOLUTION

OOI SZEE YEN

DEPARTMENT OF CHEMISTRY
FACULTY OF SCIENCE AND TECHNOLOGY
UNIVERSITY COLLEGE TERENGGANU
UNIVERSITI PUTRA MALAYSIA
2000

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KOLEJ UNIVERSITI TERENGGANU
21030 KUALA TERENGGANU

1100024824

1100024824

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KOLEJ UNIVERSITI SAINS & TEKNOLOGI MALAYSIA
(KUSTEM)

Pengarang *Ooi Szee Yen* No. Rangkaian *dn 778*

Judul *LP 16 FST 1*

Tarikh Waktu Pemulangan Nombor Ahli *2000* Tanda Tangan

*LP
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FST
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2000*

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PERPUSTAKAAN KUSTEM

TESIS

**ASSOCIATION BEHAVIOUR OF
MIXED BILE SALT AND SODIUM OCTANOATE
IN AQUEOUS SOLUTION**

By

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**Thesis Submitted in Partial Fulfillment of the Requirement for the
Degree of Bachelor of Science (Hons.)
in the Faculty of Science and Technology
University College Terengganu
University Putra Malaysia Terengganu**

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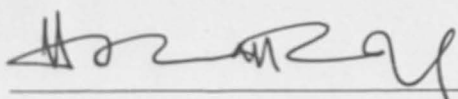
ASSOCIATION BEHAVIOR OF
MIXED BILE SALT AND SODIUM OCTANOATE
IN AQUEOUS SYSTEM

By

OOI SZEE YEN

Approved by:

Supervisor



(Prof. Dr. Hamdan B. Hj. Suhaimi)

Date: 3 May 2000

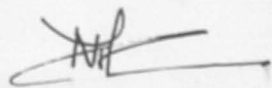
Co- Supervisor



(Dr. Mohd. Effendy ABD. Wahid)

Date: 4 May 2000

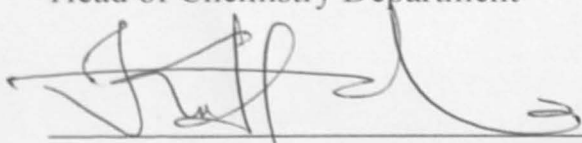
Project Co-ordinator



(Assoc. Prof. Dr. Norhayati B. Mohd. Tahir)

Date: 4 May 2000

Head of Chemistry Department



(Assoc. Prof. Dr. Ku Halim Ku Bulat)

Date: 3 Mei 2000

ABSTRACT

Studies on a mixed anionic surfactant system of bile salt and sodium octanoate/ decanol/ water system are carried out at $27 \pm 1^\circ\text{C}$. Results indicated a composition of 0.9 mole fraction of bile salt exhibit a stable and largest micellar region. Results also indicated that the mixture behaved slightly nonideally and its superior properties are due to its synergistic interaction amongst the mixed which gave a molecular interaction parameter value, β of -6 to -6.5 .

ACKNOWLEDGEMENT
ABSTRAK

Kajian terhadap sistem surfaktan garam hempedu dan sodium octanoate/ decanol/ air telah dilakukan pada suhu $27 \pm 1^\circ\text{C}$. Keputusan menunjukkan komposisi 0.9 pecahan mol garam hempedu adalah stabil dan mempunyai kawasan misel yang paling luas. Keputusan juga menunjukkan bahawa campuran tersebut menghampiri sifat tidak unggul dan sifat yang lebih baik adalah disebabkan oleh interaksi sinergi antara kedua- dua surfaktan. Nilai parameter interaksi molekul, β adalah di antara -6 ke -6.5.

I would like to thank the member of examination board, for their patience in reviewing my thesis. I am also thankful to Associate Professor Dr. Ku Hafsa Ku Bolal for his advice and guidance. For all the assistance and help from the lab assistants Muzi, Kak Hashish and Man, I would like to extend my gratitude.

All my thanks also extended to my companions at chemistry research laboratory especially to Poh Yee, Swee Khui and Yow Fai. And special thanks to Tuen Yang who has given me all the help and support throughout my research.

I cannot adequately express my gratitude to my eternal comse mate who makes these years of study a most memorable one.

Last but not least, my thanks to my family member, Papa, Mummy, Sze Lin, Sze Chen, Sze Tian and Wai Peng, who has given me their support and loving encouragement as well as tower of strength in my hour of need.